

Unit 7 Balancing Chemical Reactions Worksheet 2 Answers

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Unit 7 Chemical Equations and Reactions What is a Chemical Equation? A Chemical Equation is a written representation of the process that occurs in a chemical reaction. A chemical equation is written with the Reactants on the left side of an arrow and the Products of the chemical reaction on the right side of the equation.

Unit 7 Chemical Equations and Reactions

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Unit 7 Chemical Equations and Reactions

Unit 7: Chemical Equations Page 8 Types of Chemical Reactions Directions (a) Write and balance the given equation. (b) Indicate the type of chemical reaction represented. 1. Iron reacts with oxygen gas to produce Iron (III) oxide. (a) (b) 2. Propane (C₃H₈) reacts with oxygen gas to produce carbon dioxide and water. (a) (b) 3.

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Use mathematics and computational thinking to balance chemical reactions (i.e., synthesis, decomposition, single replacement, double replacement, and combustion) and construct an explanation for the outcome of a simple chemical

reaction based on the outermost electron states of atoms, trends in the periodic table, and knowledge of the patterns ...

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